Lists of atomic coordinates, displacement parameters, structure factors and complete geometry have been deposited with the IUCr (Reference: NA1276). Copies may be obtained through The Managing Editor, International Union of Crystallography, 5 Abbey Square, Chester CHI 2HU, England.

## References

Allen, F. H., Davies, J. E., Galloy, J. J., Johnson, O., Kennard, O., Macrae, C. F., Mitchell, E. M., Mitchell, G. F., Smith, J. M. \& Watson, D. G. (1991). J. Chem. Inf. Comput. Sci. 31, 187-204.
Allen, F. H., Kennard, O., Watson, D. G., Brammer, L., Orpen, A. G. \& Taylor, R. (1987). J. Chem. Soc. Perkin Trans. 2, pp. S1-S 19.
Domenicano, A. \& Murray-Rust, P. (1979). Tetrahedron Lett. 24, 2283-2286.
Domenicano, A., Vaciago, A. \& Coulson, C. A. (1975). Acta Cryst. B31, 221-234.
Enraf-Nonius (1989). CAD-4 Software. Version 5.0. Enraf-Nonius, Delft, The Netherlands.
Frenz, B. A. (1985). Enraf-Nonius SDP-Plus Structure Determination Package. Version 3.0. Enraf-Nonius, Delft, The Netherlands.
Hodgson, H. H. \& Beard, H. G. (1925). J. Chem. Soc. 127, 875.
Iwasaki, F. (1973). Chem. Lett. pp. 227-228.
Iwasaki, F. (1977). Acta Cryst. B33, 1646-1648.
Iwasaki, F., Tanaka, I. \& Aihara, A. (1976). Acta Cryst. B32, 1264 1266.

Johnson, C. K. (1976). ORTEPII. Report ORNL-5138. Oak Ridge National Laboratory, Tennessee, USA.
Main, P., Fiske, S. J., Hull, S. E., Lessinger, L., Germain, G., Declercq, J.-P. \& Woolfson, M. M. (1982). MULTAN11/82. A System of Computer Programs for the Automatic Solution of Crystal Structures from X-ray Diffraction Data. Universities of York, England, and Louvain, Belgium.
North, A. C. T., Phillips, D. C. \& Mathews, F. S. (1968). Acta Cryst. A24, 351-359.
Sheldrick, G. M. (1993). SHELXL93. Program for the Refinement of Crystal Structures. University of Göttingen, Germany.
Spek, A. L. (1990). Acta Cryst. A46, C-34.
Velavan, R., Sureshkumar, P., Sivakumar, K. \& Natarajan, S. (1995). Acta Cryst. C51, 1131-1133.

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# 13-Phenyl-11,13,15-triazatricyclo[8.5.2.0 $\left.{ }^{11,15}\right]$ heptadec-16-ene-12,14-dione 

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#### Abstract

The title compound, $\mathrm{C}_{20} \mathrm{H}_{25} \mathrm{~N}_{3} \mathrm{O}_{2}$, has been assigned an endo structure on the basis of its X-ray analysis.


In the $N$-phenyl-1,2,4-triazoline-3,5-dione moiety, the $\mathrm{N}-\mathrm{N}$ and average $\mathrm{C}-\mathrm{O}$ bond distances have values of 1.424 (3) and 1.208 (2) $\AA$, respectively. The closest intermolecular contact of $3.306(4) \AA$ is between a phenyl C atom and an O atom of the 1,2,4-triazoline-3,5-dione moiety.

## Comment

The Diels-Alder reaction of 1,3 -cyclooctadiene with $N$-phenyl-1,2,4-triazoline-3,5-dione (PTAD) results in a cyclo-adduct with an almost planar hydrazine moiety (Brock, Demir \& Watt, 1995). The molecule is one of the most planar of the 1,3-diene adducts of PTAD found in version 5.07 of the Cambridge Structural Database (Allen, Kennard \& Taylor, 1983). The dihedral angle between the triazoline ring and the bridging moiety increases with increasing length of the bridge (Agmon, Kaftory, Nelson \& Blakstock, 1986). By a conversion reaction of epoxide to 1,3-diene, we obtained a 1,3-cyclododecadiene. The Diels-Alder adduct with $N$-phenyl-1,2,4-triazoline-3,5-dione was identified as the title compound, but it was impossible to identify the structure as endo or exo on the basis of ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR spectral data. The present X-ray crystal structure determination resolved this ambiguity and showed it to be the endo product, i.e endo-(4).

(1)

exo-(4)
(2)

endo-(4)

In the cyclododecene part of the molecule, there is only one $\mathrm{C}=\mathrm{C}$ double bond, namely, $\mathrm{Cl0}=\mathrm{Cl1}$; all the other $\mathrm{C}-\mathrm{C}$ bond lengths have values in the singlebond range, with $\mathrm{C} 9-\mathrm{Cl} 0$ and $\mathrm{Cl1}-\mathrm{Cl} 2$ being $\mathrm{C}_{s p}{ }^{2}-$ $\mathrm{C}_{s p^{3}}$ bonds, while the rest are $\mathrm{C}_{s p^{3}}-\mathrm{C}_{s p^{3}}$ (Table 1). The $\mathrm{C} 12-\mathrm{N} 2$ and $\mathrm{C} 9-\mathrm{N} 3$ bonds are $\mathrm{C}_{s p^{3}}-\mathrm{N}_{s p^{3}}$, and are longer than the other $\mathrm{C}-\mathrm{N}$ bonds. The dihedral angle between the phenyl and triazoline-3,5-dione rings is $57.8(1)^{\circ}$. The ring $\mathrm{N} 3-\mathrm{C} 9-\mathrm{C} 10-\mathrm{Cl1}-\mathrm{C} 12-\mathrm{N} 2$ is in a boat conformation, with the C 9 and C 12 atoms -0.378 (3) and -0.607 (3) $\AA$, respectively, from the least-squares plane through atoms $\mathrm{N} 2, \mathrm{~N} 3, \mathrm{C} 10$ and C 11 .

The dihedral angle between the ring $\mathrm{N} 1-\mathrm{C} 7-\mathrm{N} 2-$ $\mathrm{N} 3-\mathrm{C} 8$ and the plane through $\mathrm{N} 2, \mathrm{C} 12, \mathrm{C} 9$ and N 3 is $29.5(2)^{\circ}$, which fits the trend described by Agmon et al. (1986). The closest intermolecular contact is 3.306 (4) $\AA$, between the $\mathrm{C} 3(-x, 1-y,-z)$ atom of the phenyl ring and the O 2 atom of the 1,2,4-triazoline-3,5dione moiety.


Fig. 1. The molecular structure of endo-(4) (ORTEPII; Johnson, 1976) with the atomic numbering scheme. The displacement ellipsoids are drawn at the $50 \%$ probability level and H atoms are shown as small circles of arbitrary radii.

## Experimental

Cyclododecadiene ( $656 \mathrm{mg}, 4 \mathrm{mmol}, 1.4$ equivalents) was dissolved in 30 ml of $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ and triazoline dione $(612.5 \mathrm{mg}$, 3.5 mmol ) was added to this solution. The mixture was stirred at room temperature for 18 h (the red colour of the reaction mixture disappeared). The solvent was evaporated and the crude product was purified by column chromatography (silica gel $60, \mathrm{EtOAc} /$ hexane, 2:1) resulting in a colourless solid ( $830 \mathrm{mg}, 70 \%$ ) which was recrystallized from EtOAc/hexane (2:1) (m.p. 435-436 K). IR (KBr): 2900-2870, 1760, 1704, 1597, 1503, 1428, $1407 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}\right): \delta 1.16-2.31$ ( $m, 15 \mathrm{H}, \mathrm{CH}_{2}$ ), 2.78-2.96 ( $m, 1 \mathrm{H}, \mathrm{CH}_{2}$ ), 4.38-4.50 ( $m, 1 \mathrm{H}$, NCH), 5.0-5.11 ( $m, 1 \mathrm{H}, \mathrm{NCH}$ ), 6.05-6.38 ( $m, 2 \mathrm{H}$, olefinic H), 7.25-7.56 ( $m, 5, \mathrm{Ar}-\mathrm{H}$ ). Analysis calculated for $\mathrm{C}_{20} \mathrm{H}_{25} \mathrm{~N}_{3} \mathrm{O}_{2}$ : C 70.76, H 7.43, N 12.38\%; found: C 70.85, H 7.38, N $12.46 \%$.

## Crystal data

$\mathrm{C}_{20} \mathrm{H}_{25} \mathrm{~N}_{3} \mathrm{O}_{2}$
Mo $K \alpha$ radiation
$M_{r}=339.441$
Orthorhombic
Pbca

$$
\begin{aligned}
& a=23.986(3) \AA \\
& b=14.076(2) \AA \\
& c=10.493(2) \AA \\
& V=3543.0(7) \AA^{3} \\
& Z=8 \\
& D_{x}=1.2726 \mathrm{Mg} \mathrm{~m}^{-3} \\
& D_{m} \text { not measured }
\end{aligned}
$$

## Data collection

Enraf-Nonius CAD-4 diffractometer
$\lambda=0.71069 \AA$
Cell parameters from 25 reflections
$\theta=10.20-18.11^{\circ}$
$\mu=0.078 \mathrm{~mm}^{-1}$
$T=295 \mathrm{~K}$
Prism
$0.50 \times 0.40 \times 0.35 \mathrm{~mm}$
Colourless

2063 reflections with $I>\sigma(I)$
$\omega / 2 \theta$ scans
Absorption correction: empirical via $\psi$ scans (MolEN; Fair, 1990) $T_{\text {min }}=0.961, T_{\text {max }}=0.973$
3784 measured reflections 3700 independent reflections
$R_{\text {int }}=0.012$
$\theta_{\text {max }}=25.7^{\circ}$
$h=-29 \rightarrow 0$
$k=0 \rightarrow 17$
$l=0 \rightarrow 12$
3 standard reflections frequency: 120 min intensity decay: $-1.02 \%$

## Refinement

Refinement on $F$
$R=0.052$
$w R=0.056$
$S=0.86$
2063 reflections
226 parameters
H atoms: see text
$(\Delta / \sigma)_{\max }=0.00018$
$\Delta \rho_{\text {max }}=0.234 \mathrm{e}^{\AA^{-3}}$
$\Delta \rho_{\text {min }}=-0.118 \mathrm{e}^{-3}$
Extinction correction: none
Scattering factors from International Tables for $X$-ray Crystallography (Vol. IV)

Table 1. Selected geometric parameters $\left(\AA^{\circ}{ }^{\circ}\right)$

| $\mathrm{O} 1-\mathrm{C} 7$ | $1.206(3)$ | $\mathrm{C} 9-\mathrm{C} 20$ | $1.537(5)$ |
| :--- | :--- | :--- | :--- |
| $\mathrm{O} 2-\mathrm{C} 8$ | $1.211(3)$ | $\mathrm{C} 10-\mathrm{C} 11$ | $1.312(4)$ |
| $\mathrm{N} 1-\mathrm{C} 1$ | $1.430(4)$ | $\mathrm{C} 11-\mathrm{C} 12$ | $1.497(4)$ |
| $\mathrm{N} 1-\mathrm{C} 7$ | $1.405(3)$ | $\mathrm{C} 12-\mathrm{C} 13$ | $1.518(4)$ |
| $\mathrm{N} 1-\mathrm{C} 8$ | $1.387(4)$ | $\mathrm{C} 13-\mathrm{C} 14$ | $1.526(4)$ |
| $\mathrm{N} 2-\mathrm{N} 3$ | $1.424(3)$ | $\mathrm{C} 14-\mathrm{C} 15$ | $1.516(5)$ |
| $\mathrm{N} 2-\mathrm{C} 7$ | $1.357(4)$ | $\mathrm{C} 15-\mathrm{C} 16$ | $1.515(5)$ |
| $\mathrm{N} 2-\mathrm{C} 12$ | $1.470(4)$ | $\mathrm{C} 16-\mathrm{C} 17$ | $1.522(5)$ |
| $\mathrm{N} 3-\mathrm{C} 8$ | $1.377(4)$ | $\mathrm{C} 17-\mathrm{C} 18$ | $1.532(5)$ |
| $\mathrm{N} 3-\mathrm{C} 9$ | $1.495(4)$ | $\mathrm{C} 18-\mathrm{C} 19$ | $1.536(5)$ |
| $\mathrm{C} 9-\mathrm{C} 10$ | $1.497(5)$ | $\mathrm{C} 19-\mathrm{C} 20$ | $1.527(5)$ |
| $\mathrm{C} 1-\mathrm{N} 1-\mathrm{C} 7$ | $124.8(2)$ | $\mathrm{C} 9-\mathrm{C} 10-\mathrm{C} 11$ | $121.3(3)$ |
| $\mathrm{C} 1-\mathrm{N} 1-\mathrm{C} 8$ | $123.9(2)$ | $\mathrm{C} 10-\mathrm{C} 11-\mathrm{C} 12$ | $117.2(3)$ |
| $\mathrm{C} 7-\mathrm{N} 1-\mathrm{C} 8$ | $111.0(2)$ | $\mathrm{N} 2-\mathrm{C} 12-\mathrm{C} 11$ | $108.1(2)$ |
| $\mathrm{N} 3-\mathrm{N} 2-\mathrm{C} 7$ | $109.9(2)$ | $\mathrm{N} 2-\mathrm{C} 12-\mathrm{C} 13$ | $114.9(2)$ |
| $\mathrm{N} 3-\mathrm{N} 2-\mathrm{C} 12$ | $114.1(2)$ | $\mathrm{C} 11-\mathrm{C} 12-\mathrm{C} 13$ | $116.9(3)$ |
| $\mathrm{C} 7-\mathrm{N} 2-\mathrm{C} 12$ | $130.6(2)$ | $\mathrm{C} 12-\mathrm{C} 13-\mathrm{C} 14$ | $115.0(3)$ |
| $\mathrm{N} 2-\mathrm{N} 3-\mathrm{C} 8$ | $107.5(2)$ | $\mathrm{C} 13-\mathrm{C} 14-\mathrm{C} 15$ | $116.1(3)$ |
| $\mathrm{N} 2-\mathrm{N} 3-\mathrm{C} 9$ | $118.9(2)$ | $\mathrm{C} 14-\mathrm{C} 15-\mathrm{C} 16$ | $117.2(3)$ |
| $\mathrm{N} 1-\mathrm{C} 7-\mathrm{N} 2$ | $104.9(2)$ | $\mathrm{C} 15-\mathrm{C} 16-\mathrm{C} 17$ | $113.8(3)$ |
| $\mathrm{N} 1-\mathrm{C} 8-\mathrm{N} 3$ | $105.9(2)$ | $\mathrm{C} 16-\mathrm{C} 17-\mathrm{C} 18$ | $116.7(3)$ |
| $\mathrm{N} 3-\mathrm{C} 9-\mathrm{C} 10$ | $109.5(3)$ | $\mathrm{C} 17-\mathrm{C} 18-\mathrm{C} 19$ | $115.1(3)$ |
| $\mathrm{N} 3-\mathrm{C} 9-\mathrm{C} 20$ | $114.4(3)$ | $\mathrm{C} 18-\mathrm{C} 19-\mathrm{C} 20$ | $115.1(3)$ |
| $\mathrm{C} 10-\mathrm{C} 9-\mathrm{C} 20$ | $115.4(3)$ | $\mathrm{C} 9-\mathrm{C} 20-\mathrm{C} 19$ | $119.6(3)$ |

H atoms were placed geometrically $0.95 \AA$ from their parent C atoms. The H atoms of cyclododecadiene were refined for a few cycles and then a riding model was used for all H atoms, with $U_{\mathrm{eq}}(\mathrm{H})=1.3 U_{\mathrm{eq}}(\mathrm{C})$.

Data collection: CAD-4 Express (Enraf-Nonius, 1993). Data reduction: MolEN (Fair, 1990). Program(s) used to solve structure: $S I R$ in MolEN. Program(s) used to refine structure: LSFM in MolEN. Molecular graphics: PLATON92 (Spek, 1992) version of ORTEP (Johnson, 1976). Software used to prepare material for publication: MolEN.

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Lists of atomic coordinates, displacement parameters, structure factors and complete geometry have been deposited with the $I \mathrm{IUCr}$ (Reference: AB1419). Copies may be obtained through The Managing Editor, International Union of Crystallography, 5 Abbey Square, Chester CHI 2HU, England.

## References

Agmon, I., Kaftory, M., Nelson, S. F. \& Blakstock, S. C. (1986). J. Am. Chem. Soc. 108, 4477-4484.
Allen, F. H., Kennard, O. \& Taylor, R. (1983). Acc. Chem. Res. 16, 146-153.
Brock, C. P., Demir, A. S. \& Watt, D. S. (1995). Acta Cryst. C51, 2434-2437.
Enraf-Nonius (1993). CAD-4 Express Software. Version 1.1. EnrafNonius, Delft, The Netherlands.
Fair, C. K. (1990). MolEN. An Interactive Intelligent System for Crystal Structure Analysis. Enraf-Nonius, Delft, The Netherlands.
Johnson, C. K. (1965). ORTEP. Report ORNL-3794. Oak Ridge National Laboratory, Tennessee, USA.
Spek, A. L. (1992). PLUTON92. Molecular Graphics Program. University of Utrecht, The Netherlands.

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## 4,4'-Dimethyl-2,2'-bipyridine 1,1'-Dioxide

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## Abstract

There are four molecules in a unit cell of the title compound, $\mathrm{C}_{12} \mathrm{H}_{12} \mathrm{~N}_{2} \mathrm{O}_{2}$, and the structure of only half a molecule has been determined because of its twofold rotational symmetry. The dihedral angle between the two pyridine rings is $61.3^{\circ}$ and the non-bonded distance between O atoms in the $N$-oxide groups is $2.96 \AA$. The skew conformation of the skeleton is almost the same as that of the tetrakis( $2,2^{\prime}$-bipyridine $1,1^{\prime}$-dioxide)lanthanum(III) complex.

## Comment

The bidentate ligand 2,2'-bipyridine $1,1^{\prime}$-dioxide (bpdo) forms a skewed seven-membered ring with a metal ion. Since the skew conformation produces a pair of enantiomers, a tris(bpdo) complex has four possible racemic pairs of conformational isomers. The tris-(bpdo)- $\mathrm{Cr}^{\text {III }}$ complex, however, crystallizes in one of the four isomers and then isomerizes and racemizes on dissolution in water (Kanno \& Fujita, 1987). This
indicates that the skewed bpdo-chelate ring is flexible and changes its conformation rather easily in solution. The dihedral angle between the pyridine rings, which seems to represent a distortion of the seven-membered ring, is $61.4^{\circ}$ in the $\mathrm{La}^{\text {III }}$ complex (Al-Karaghouli, Day \& Wood, 1978) and $53.7^{\circ}$ in the $\mathrm{U}^{\mathrm{IV}}$ complex (Alcock \& Roberts, 1987). The present study was undertaken in order to estimate the dihedral angle between the pyridine rings of the bpdo molecule with methyl groups at the 4,4'-positions, (I).

(I)

The title molecule has a twofold rotational axis and the parameters of half of the molecule have been determined (Fig. 1). The pyridine ring is planar within $0.03 \AA$, but the O atom of the $N$-oxide group is slightly out of the plane $(0.13 \AA)$, away from the other $O$ atom


Fig. 1. ORTEP drawing (Johnson, 1965) of the whole title molecule, half of which is an asymmetric unit, showing the atomic numbering scheme. Displacement ellipsoids of the non-H atoms are shown at the $50 \%$ probability level. H atoms are shown as spheres of arbitrary radii.


Fig. 2. The crystal structure of (I) viewed along the $b$ axis.

